

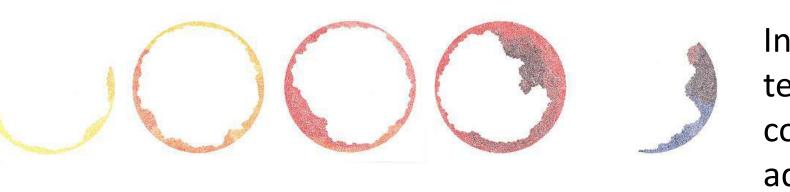


Modern inks. investigation of felt-tip pens

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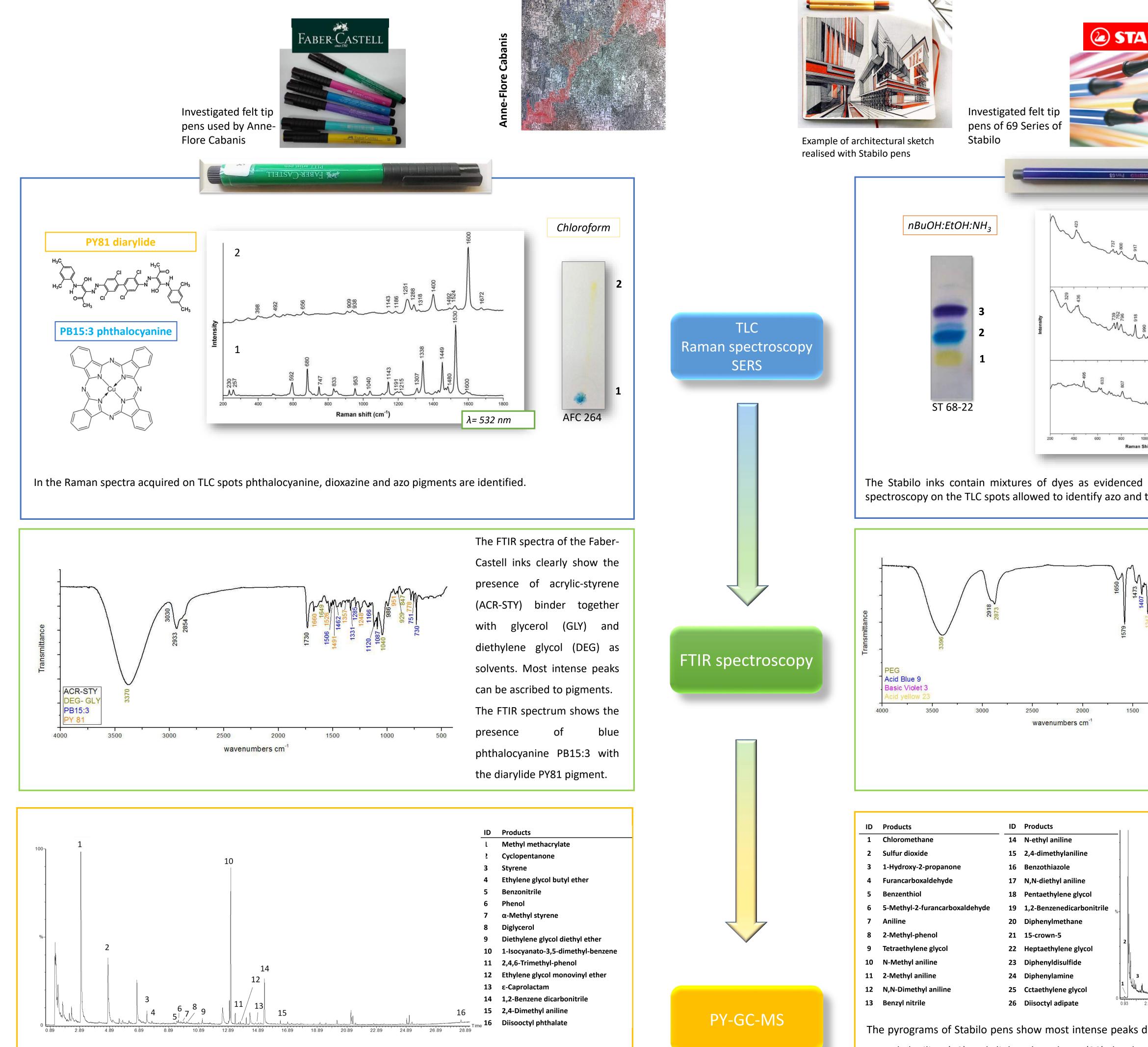
Couleurs



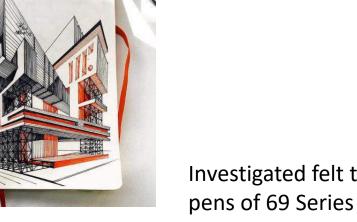
Inks of felt-tip pens are used by modern artists for the realization of sketches, drawings, copies, architectural drawings and other technical designs. However, these inks are usually very sensitive to light and chemical agents and the exact knowledge of their composition may be important to define the optimal conservation treatment and storage conditions. So far, few studies have been addressed to the chemical characterisation of these materials [1] and often information on binders, fillers, dyes and pigments is lacking.

In this study felt tip pens (Faber-Castell) used by the French modern artist Anne-Flore Cabanis and other artist pens (Stabilo) were investigated by using an integrated analytical approach. The inks have been analysed with Fourier transform infrared (FT-IR), μ -Raman and visible reflectance spectroscopy, and pyrolysis - gas chromatography - mass spectrometry (Py-GC-MS).

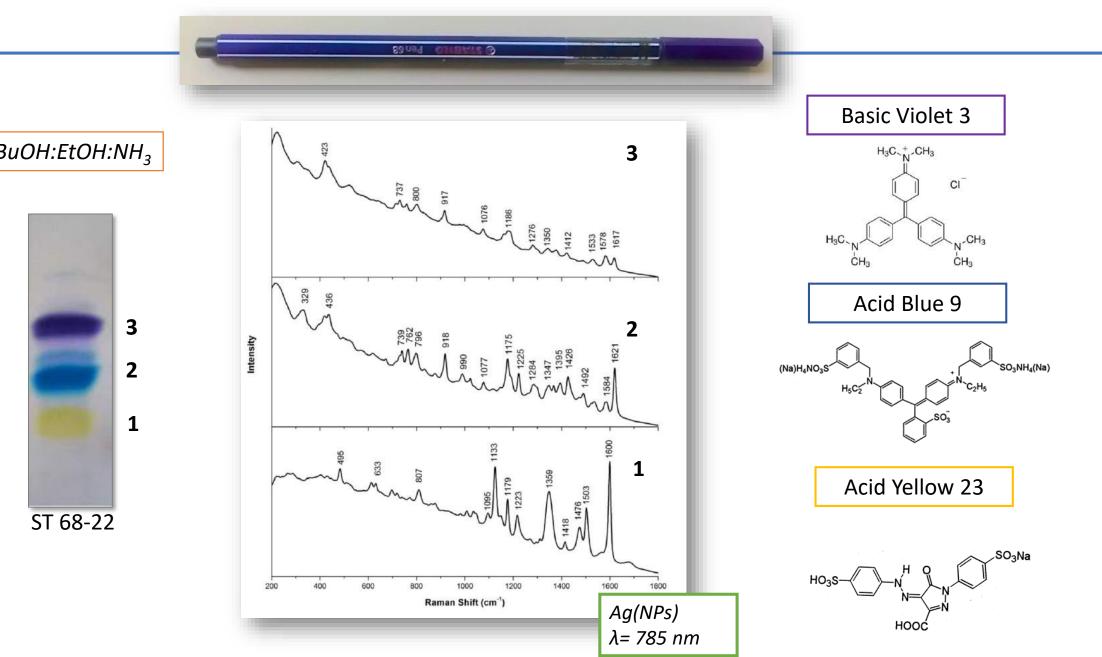
Sanse Titre



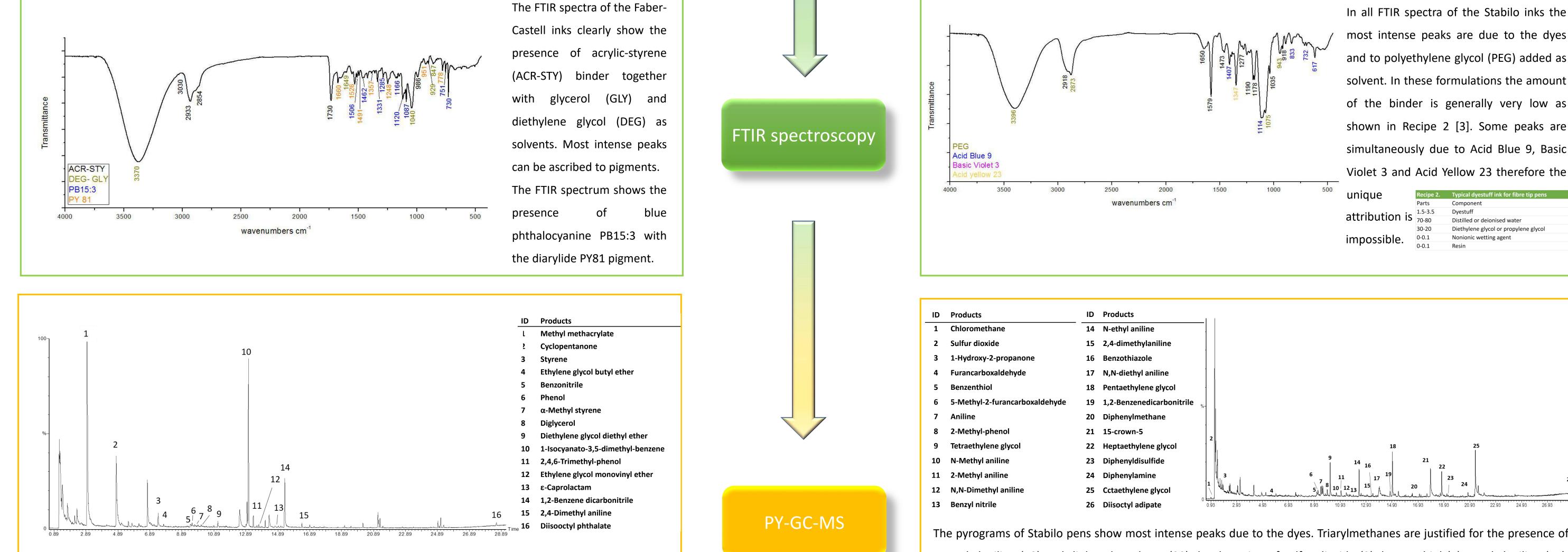








The Stabilo inks contain mixtures of dyes as evidenced by thin layer chromatography (TLC). Surface-enhanced Raman spectroscopy on the TLC spots allowed to identify azo and triarylmethane dyes [2], and rhodamine B.



In all FTIR spectra of the Stabilo inks the most intense peaks are due to the dyes and to polyethylene glycol (PEG) added as solvent. In these formulations the amount

Parts

1.5-3.5

30-20

21

16.93

14.93

22

23

18.93 20.93

22.93

24.93

0-0.1

Typical dyestuff ink for fibre tip pens

Diethylene glycol or propylene glycol

Distilled or deionised water

Nonionic wetting agent

Componen

Dvestuff

In all pyrograms of the Faber-Castell inks a methyl methacrylate-styrene copolymer binder was identified. The presence of benzonitrile (5) and 1,2 benzenedicarbonitrile (14) suggests the addition of PB15:3; whereas the azo PY81 pigment may be confirmed by the occurence of 1-Isocyanato-3,5-dimethyl-benzene (10) and 2,4-dimethyl aniline (15) [5]. Diethylen glycol (9) and glycerol are used as solvent; phenol and phthalates are added respectively as antioxidant and plasticizer.

The pyrograms of Stabilo pens show most intense peaks due to the dyes. Triarylmethanes are justified for the presence of N-methylaniline (10) and diphenyl methane (20). The detection of sulfur dioxide (2), benzenthiol (5), N-ethylaniline (14), and N,N-diethyl aniline (16) can be related to Acid Blue 9 (C.I. 42090); while chloromethane (1), 2,4-dimethyl aniline (15), and N,N-dimethyl aniline (12) point to the occurrence of **Basic Violet 3** (C.I. 42555). Pyrolysis products such as benzyl nitrile (13), benzothiazole (16), 1,2-benzenedicarbonitrile (19), and diphenyldisulfide (h) confirm the addition of Acid Yellow 23 (C.I. 19140). In these formulations traces of 1-hydroxy-2-propanone (3), furancarboxaldehyde (4) and 5-methylfurancarboxaldehyde (6) suggest the presence of plant gum as binder [4]. Polyethylene glycol (pyrolysis products: 9, 18, 21, 22 and 25) has been added as a solvent and phenol (8) as antioxidant.

 $t_0 t_3$

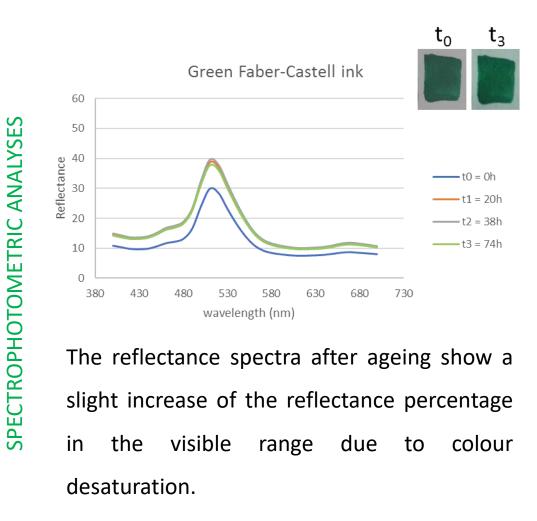
------ t0 = 0h

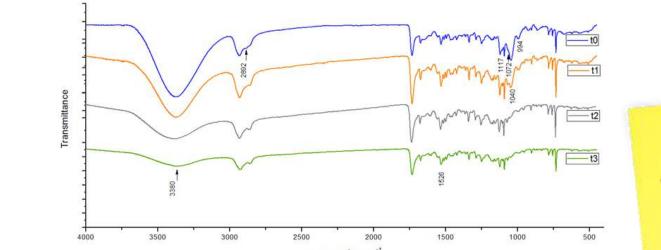
_____ t1 = 20h

------ t2 = 38h

—— t3 = 74h

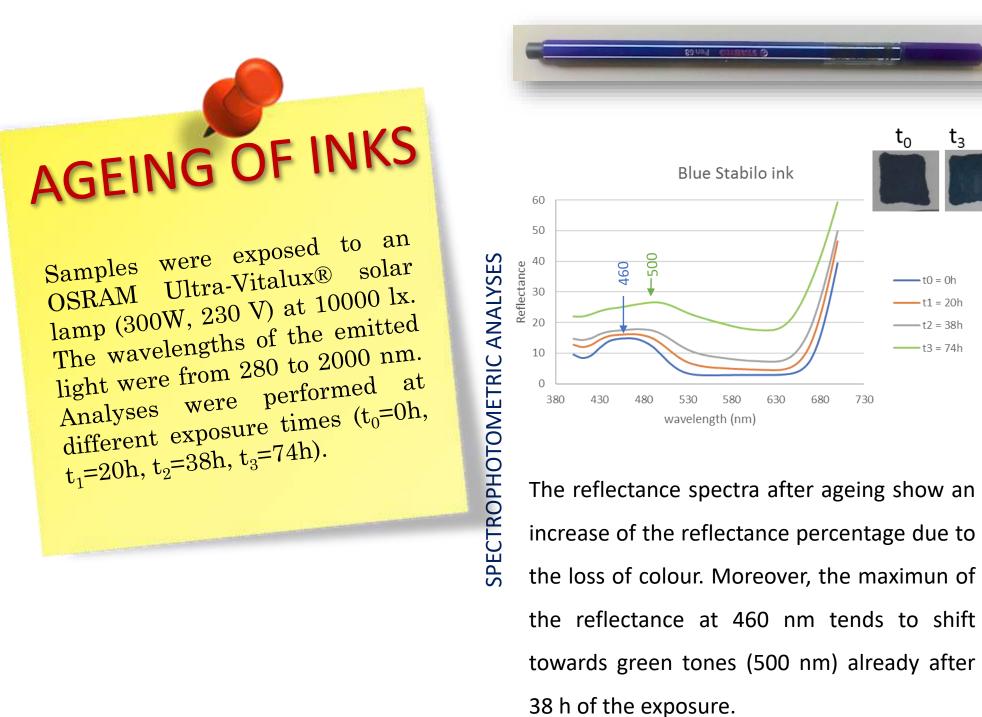




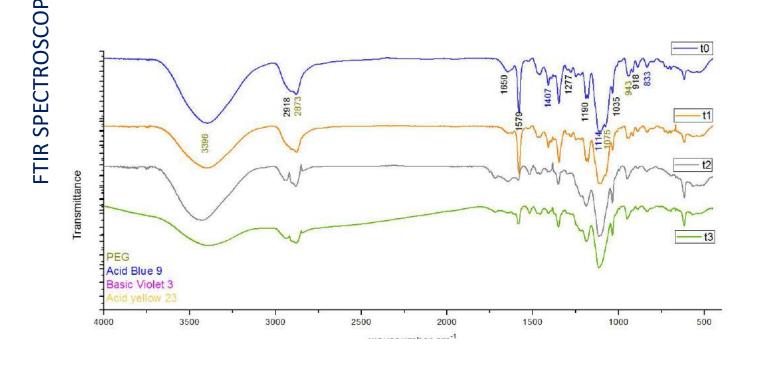


FTIR spectra acquired before and after different exposure times show some differences mainly due to the evaporation of the solvents confirming that pigment lightfastness and the type of binder show relatively high photo-stability. The bands at 3370 cm⁻¹ (GLY-DEG), at 1040 cm⁻¹ (GLY), at 994 and 847 cm⁻¹ (DEG) had noticeably decreased in intensity after 38h of exposure; at the same time, the absorptions due to the PB 15:3 (1120 and 1087 cm⁻¹) became more clearly evident. The absorptions related to pigments and

binders seem to be stable under these aging conditions.



FTIR spectra of Stabilo inks after aging show the reduction of peaks attributed to PEG (3396, 2873, 1075 and 943 cm⁻¹). Absorptions at 1650, 1579, 1277, 1190, 1035 and 918 cm⁻¹ ascribed to triarylmethanes (Acid Blue 9 and Basic Violet 3) and at 1407 and 1114 cm⁻¹ uniquely due to Acid Blue 9, reduce noticeably their intensities over time. Acid yellow 23 (1347 cm⁻¹) instead, shows higher resistance to photodegradation.



References

[1] F. C. Izzo, et al. Microchem. J. (2016) 124, 919–928; [2] B. Doherty et al. Spectrochim Acta A Mol Biomol Spectrosc (2014) 121, 292–305; [3] G. Pfingstag, JSDC (1993) 109, 188-192; [4] O. Chiantore *et al.*, Int. J. Mass Spect. (2009) 284, 35–41

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